

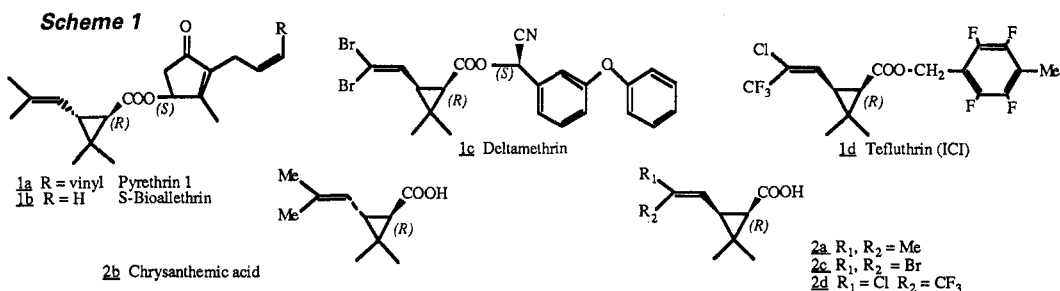
From grape to grape: novel stereoselective syntheses of chiral pyrethroids – synthesis of the most potent commercially available insecticides

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ABSTRACT. We report novel stereo- and enantioselective routes to vinyl cyclopropane carboxylic esters.

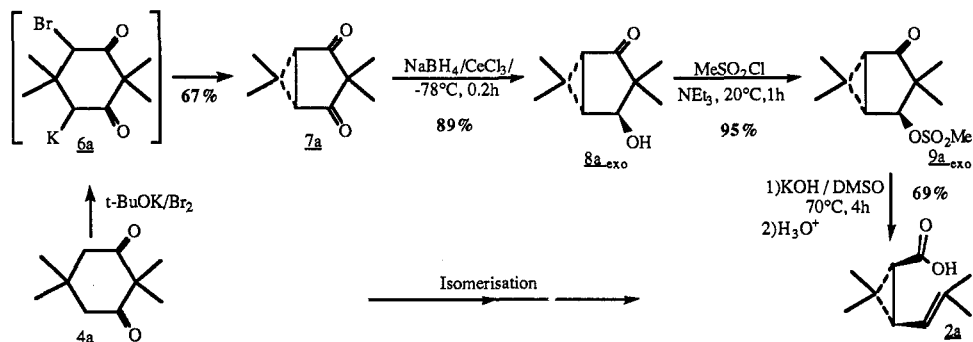
S-bioallethrin **1a,b**, deltamethrin **1a-d** **1c** and tefluthrin **1e,f** **1d** are valuable insecticides, commercially available for domestic and agricultural uses against flying and soil insects respectively. These are unnatural esters of 2,2-dimethyl-3-vinyl cyclopropane carboxylic acids **2** which all belong to the (1R)-series. But whereas the former derivative possesses the *trans*-stereochemistry found in the natural pyrethrin I **1a**, the others possess a *cis*-relationship. The structure of **2d** is clearly the most complex since it possesses in addition to the highly substituted cyclopropane ring with the *cis*-stereochemistry, a highly functionalized (E)-carbon-carbon double bond. In fact it does not exist an enantio- or a stereoselective synthesis of such a compound.^{1e}

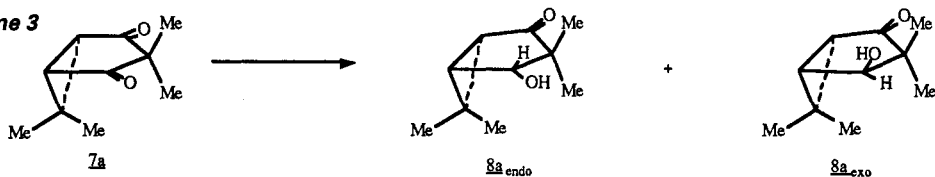


We have designed as a model a novel stereoselective synthesis of (d,l)-*cis*-chrysanthemic acid **2a** which uses cheap reagents compatible with industrial requirements and which we hope can be further applied to the enantio- and stereoselective synthesis of 1(R)-**2d**.

2,2,5,5-Tetramethyl cyclohexa-1,3-dione **2a**, readily prepared by di-methylation of dimedone **3a** (4 equiv. K_2CO_3 , EtOH-H₂O, 2.5 equiv. MeI, 70°C, 4h, 63% yield), was chosen as the starting material since it possesses not only the same formula as chrysanthemic acid **2a** ($\text{C}_{10}\text{H}_{16}\text{O}_2$) but also the carbon framework and the functionalities placed in suitable positions to allow in a minimum of steps the functional group modifications required for the desired transformation. The key steps of this process are without doubt (i) the cyclopropanation reaction which produces the bicyclo [3.1.0] hexa-2,4-dione **3a** and (ii) the Grob fragmentation ⁴ which was achieved on the mesylate **2a** resulting from its reduction-sulfonation. The stereochemical outcome of each individual step proved as expected particularly important for the success of the whole process shown in the Scheme 2.

Scheme 2



Scheme 3

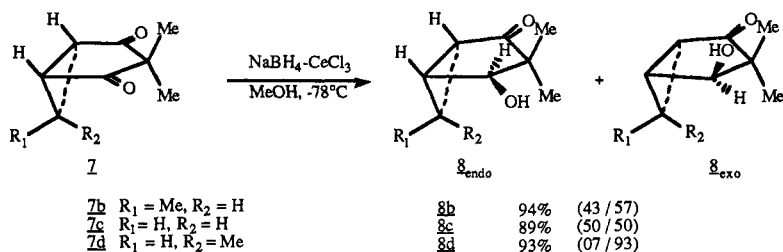
Reagent	Conditions	Overall Yield	Ratio 8a/8b
1mol. equiv. NaBH ₄	Methanol, 20°C	89%	80/20
1mol. equiv. NaBH ₄ / 1mol. equiv. CeCl ₃	Methanol, 20°C	88%	22/78
1mol. equiv. NaBH ₄ / 1mol. equiv. CeCl ₃	Methanol, -78°C	85%	0/100
1mol. equiv. NaBH ₄ / 0.5 mol. equiv. CeCl ₃	Methanol, -78°C	81%	23/77

The synthesis of the 2,2,5,5-tetramethyl-bicyclo [3.1.0] hexa-2,4-dione **7a** was efficiently achieved (67% yield) from 2,2,5,5-tetramethyl-cyclohexa-1,3-dione **4a** on sequential reaction with potassium tert-butoxide (2.2 equiv. in THF from -78°C to +40°C) and bromine (1.6 equiv. in pentane, +40°C, 1h). This one pot transformation offers an original solution to our problem since the 6-bromo-4-potassio-2,2,5,5-tetramethyl-cyclohexa-1,3-dione **6a** intermediary formed is immediately cyclized under these conditions rather than to further react with bromine. This avoids the difficulties we encountered when the transformation was performed stepwise.³

The chemoselective mono-reduction of this bicyclic dione was achieved by a large array of hydrides^{5,6} (Scheme 3) but most of the reactions delivered exclusively or mainly the endo alcohol **8a-endo** resulting from the attack of the bicyclic diketone from the less hindered face. Unfortunately however this alcohol proved to be unsuitable for further transformation since it is too hindered and none of the conditions which we tried allowed its transformation to the corresponding sulfonate, the starting material on which Grob fragmentation is expected to take place.⁴

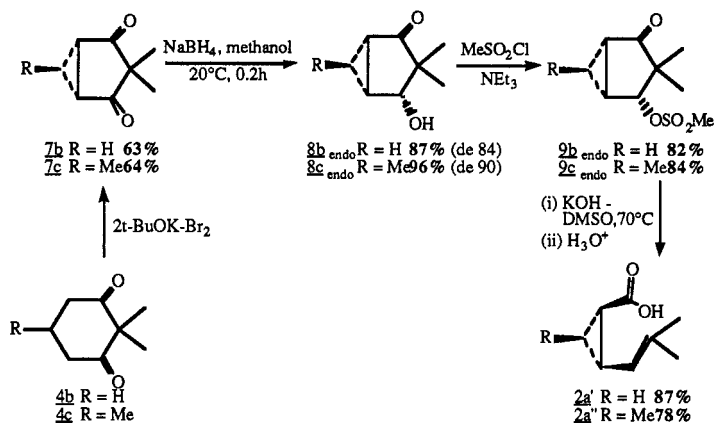
After different unsuccessful essays we have found that a dramatic change in the exo/endo ratio was achieved when the reaction is carried out with NaBH₄ in methanol but in the presence of cerium trichloride.^{7,8} Interestingly an exclusive endo attack leading to **8a-exo** was observed when the reaction is carried out at low temperature (-78°C). This reversal of regioselectivity was accounted from a pre-complexation by CeCl₃ of the bicyclic diketone **7a** from its less hindered exo-face and reaction of the hydride from the endo face of the complexed diketone. Therefore the presence of the endo methyl group should be crucial for the stereoselective reduction since its role is to hinder the endo-face allowing thus a more complete discrimination between the two faces of the bicyclic-diketone **7a** by CeCl₃.

We have in fact found, in accordance with this hypothesis, that (i) the reduction with NaBH₄-CeCl₃ of the bicyclic diketones **7b** and **7c** missing the endo methyl group is less selective than the previous one and (ii) a very high selectivity related to the one found in **7a** and leading to the exo-alcohol **8d-exo** is observed when the bicyclic diketone **7d** possessing an endo methyl group is reacted with the same reducing system (1 equiv. NaBH₄, 1 equiv. CeCl₃, MeOH, -78°C) (Scheme 4).

Scheme 4

The synthesis of (d,l)-*cis*-chrysanthemic acid was achieved (Scheme 2) from the exo alcohol **8a-exo** in two steps which involve its mesylation (MesCl, Net₃, CH₂Cl₂, 20°C, 1h, 95% yield) and the reaction of the resulting mesylate **9a-exo** with potassium hydroxide in aqueous DMSO (70°C, 4h, 69% yield).

Scheme 5



The whole process described in Scheme 5, carried out on the cyclohexanediones $4b$ and $4c$ missing at least one methyl group in the sixth position proceeds slightly differently than the one depicted in the scheme 2. The cyclopropanation reaction carried out on $4b$ or $4c$ works as on $4a$ [(i) 2 equiv. *t*-BuOK, THF -78°C to 20°C , (ii) Br_2 in pentane). Unexpectedly however, a perfect control of the stereochemistry leading exclusively to the bicyclic-diketone $7c$ bearing an exo-methyl group on the cyclopropane ring was achieved from $4c$. The resulting bicyclic diketones $7b$ and $7c$ have been as expected reduced mainly from their exo face with sodium borohydride in methanol but on the contrary to what was reported for $8a_{endo}$, the resulting endo alcohols $8b_{endo}$ and $8c_{endo}$ proved to be efficiently transformed to their corresponding endo mesylates $9b_{endo}$ and $9c_{endo}$ on reaction with mesyl chloride (1.1 equiv. MesCl, 1.1 equiv. NEt_3 , CH_2Cl_2 , -8 to 20°C , 1h, 83% yield each). Finally, the resulting β -ketosulfonates proved to be prone to fragment to 2,2-di-desmethyl- $2b$ or to 2-desmethyl- $2c$ chrysanthemic acids on further reaction with KOH in aqueous DMSO (6 equiv. KOH- H_2O (8-2), DMSO, $+70^\circ\text{C}$), although the *trans*-antiperiplanar arrangement required for such reaction to take place is not apparently fully achieved.

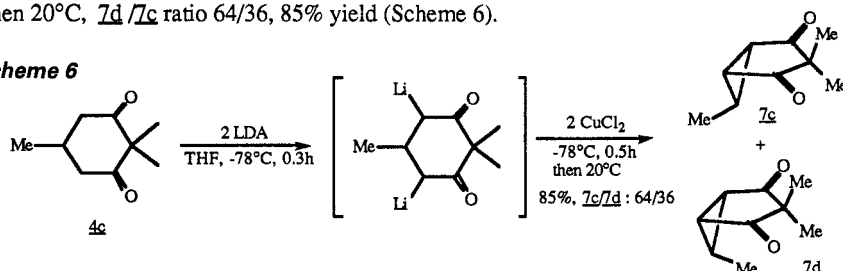
That the exo-mesylates 9 are more reactive in the Grob fragmentation reaction than their endo-stereoisomers was clearly demonstrated by the difference of reactivity observed between each of the two stereoisomers when the reaction is performed under the above mentioned conditions but on a 1 to 1 mixture of exo and endo mesylates $9b$ and $9c$ derived from the reduction of the bicyclic-diketones $7b$ and $7c$ with $\text{NaBH}_4\text{-CeCl}_3$.

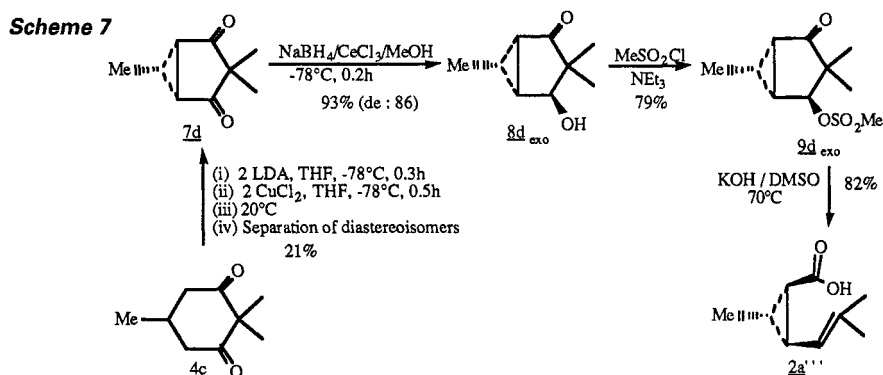
The whole process applied to the diketone $4c$ exclusively leads to the 2-desmethyl chrysanthemic acid $2a''$ which bears all the substituents *cis* one to each others (Scheme 5). The synthesis of its epimer at C-2 was more difficult since we have been unable to synthesize stereoselectively the bicyclic diketone $7d$ bearing an endo-methyl group in the sixth position.

We have in fact generated the two stereoisomers of the 4-bromo-2,2,5-trimethyl-cyclohexa-1,3-diones $5c$ and $5d$ by bromination of the trimethyl-cyclohexane dione $4c$, but they surprisingly both provide stereoselectively and exclusively the 2,2,5-trimethyl-bicyclo [3.1.0] hexa-2,4-dione $7c$ possessing the exo-methyl group on further reaction with potassium tert-butoxide (THF, -78°C , 2h).

We have been finally able to synthesize the 2,2,5-trimethyl-bicyclo [3.1.0] hexa-2,4-dione $7d$ but mixed with $7c$ by performing the cyclopropanation reaction by oxidation of the dilithio salt of the 2,2,5-trimethyl-cyclohexa-1,3-dione $4c$ with copper (II) chloride (2 equiv. LDA, -78°C , 0.5h (ii) CuCl_2 , -78°C , 0.5h, then 20°C , $7d/7c$ ratio 64/36, 85% yield (Scheme 6).

Scheme 6

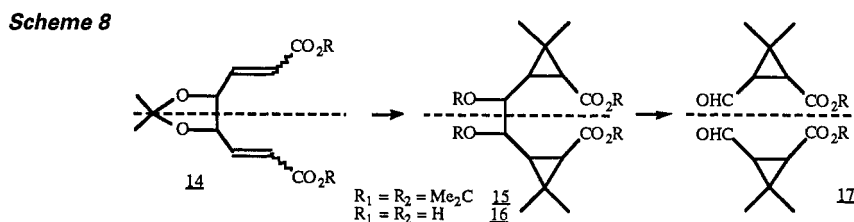




The bicyclic-diketone **7d** has been separated from **7c** by column chromatography on silicagel and the 2-desmethyl chrysanthemic acid **2a'''** has been stereoselectively synthesized from **7d** by a sequence of reactions identical to the one used for the synthesis of (d,l)-*cis*-chrysanthemic acid **2a** [(i) NaBH₄-CeCl₃, -78°C, 93% yield; (ii) MesCl, triethylamine, 20°C, 79% yield; (iii) 6 equiv. KOH, DMSO-H₂O (8-2), 70°C, 82% yield] (Scheme 7).

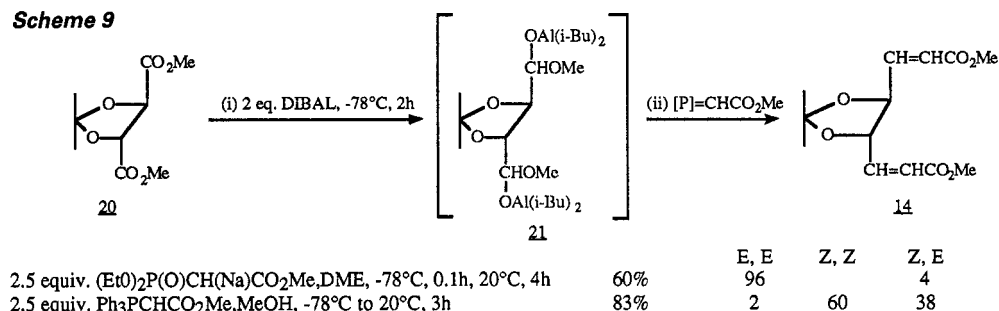
We have also investigated another approach to *trans*-**2a** and *cis*-**2b**-chrysanthemic acid and to its *cis*-dibromovinyl analogue **2c** (Scheme 1). This takes advantage of the easy cyclopropanation of suitably functionalized α,β -unsaturated esters with phosphorus and sulfur ylides and found a clear cut difference of reactivity between isopropylidetriphenylphosphorane and -diphenylsulfurane. The former reacts with (*Z*)- and (*E*)- α,β -unsaturated esters and leads exclusively to *trans*-cyclopropane carboxylic esters. It enolises γ -butyrolactone **10** and 4-methoxy- γ -butyrolactone **11** and affords ^{12b} dimethyl caronate **13** from dimethyl fumarate **12**, in high yield and with reasonably high asymmetric induction. Isopropylidenediphenylsulfurane reacts stereospecifically with α,β -unsaturated esters and leads to *trans*-cyclopropyl esters from (*E*)- α,β -unsaturated esters and to *cis*-cyclopropyl derivatives from their (*Z*)-isomers.^{14a} It provides bicyclo [3.1.0] hexane derivatives on reaction with γ -butyrolactones but affords ^{12b} dimethyl caronate **13** from dimethyl fumarate **12**, in high yield but with dramatically low asymmetric induction.

As a continuation of this work, we became interested in performing the above mentioned cyclopropanation reaction on the diesters **14** (Scheme 8). This yet unknown building block should possess exceptional features due not only to the presence of two γ -alkoxy- α,β -unsaturated ester moieties (analogous to masked γ -oxobutenates) but also due to the chiral nature of some of its forms susceptible to allow asymmetric inductions (*inter alia* in the double cyclopropanation reaction envisaged). Further hydrolysis of the dioxolane moiety in **15** was expected to produce the diol **16** precursor of the hemicaldehyde **17** the well known key intermediate in various syntheses of chrysanthemic acid **2**. Last but not least we discovered that both the (*E,E*)-**14a** and the (*Z,Z*)-**14b** stereoisomers in each series [(*4R,5R*) and (*4S,5S*)] are available from natural mannitol **18** and from natural tartaric acid **19** respectively via their acetonides.



We found (Scheme 9) that the *trans*-**14a** and *cis*-**14b** α,β -unsaturated esters can be produced in one pot from the diester **20** derived from natural (2*R*,3*R*)-(+)-tartaric acid **19** by sequential treatment with diisobutyl aluminum hydride (DIBAL, 2 equiv. of a 1.5*N* solution in toluene, -78°C, 2h) followed by reaction with sodio carbomethoxymethyl diethylphosphonate **16a** [2.5 equiv., DME, -78°C to 20°C, 4h, 60% yield of the (*E,E*)-stereoisomer **14a** containing 4% of the (*E,Z*)-isomer **14c**] or with methyl(triphenylphosphoranilidene)acetate **16b** [2.5 equiv. MeOH, -78°C to +20°C, 3h; 44% yield of the (*Z,Z*) isomer **14b**].

Scheme 9

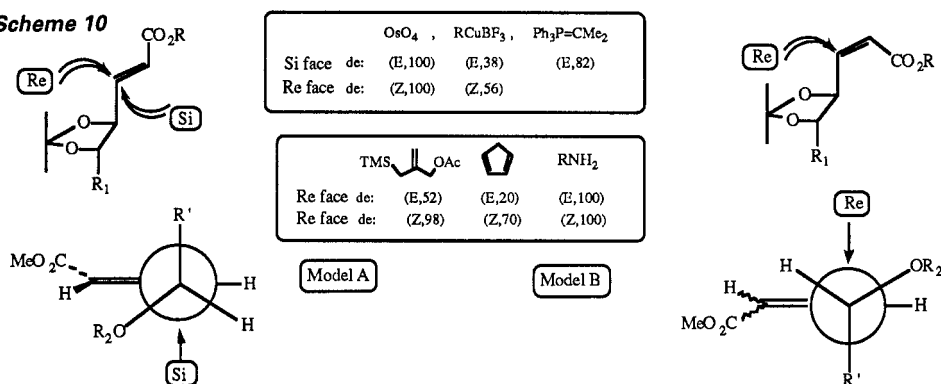


Remarkably, these Wittig-Horner or Wittig reactions presumably take place on the dialuminate **21** rather than on the corresponding dialdehyde **22** and therefore this approach avoids the tedious isolation of such highly water soluble and enolisable compound. It should also allow the synthesis of their enantiomers **14'** from the still cheap unnatural tartaric acid.

Diastereoselective addition to γ -alkoxy- α,β -unsaturated carbonyl compounds is well documented.¹⁷ It has been used *inter alia* for the enantioselective synthesis of the β -lactame thienamycin,^{17a} brefeldin,^{17b} santalene,^{17c} prostanoids,^{17c} galactitol^{17d} and (1R)-*trans*-chrysanthemic acid^{17e} **2b**.

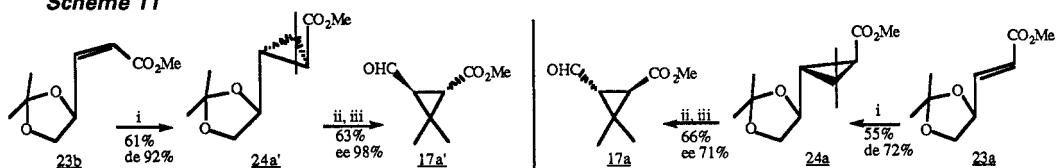
In most of the cases the stereochemistry of the β -carbon on the adduct only depends upon the stereochemistry of the γ -carbon and proved to be ^{17a-c} independent of the stereochemistry of the carbon-carbon double bond (Scheme 10). In few cases however the stereochemistry of the β -carbon on the adduct is not only dependent upon the stereochemistry of the γ -carbon but is also directly related to the stereochemistry of carbon-carbon double bonding the starting material ^{17f,h} (Scheme 10). Interestingly the stereochemical outcome on the β -carbon of the adduct, of all these reactions was found to be identical in both series when the (Z)-stereoisomer is involved. The models used to rationalize these results however remain unsatisfactory (Scheme 10).

Scheme 10



We first investigated the more simple case of unsaturated esters **23a** and **23b** derived from D-glyceraldehyde and possessing respectively the (E)- and (Z)-stereochemistry ¹⁶ (Scheme 11 and 12). When reacted with isopropylidetriphenylphosphorane, the former product leads, as previously described ^{17e} to the methyl (1R)-*trans*-cyclopropane carboxylate **24a** as a 91/9 mixture of diastereoisomers in which **24a** predominates, whereas its (1S)-diastereoisomer **24a'** still possessing the *trans*-stereochemistry is almost exclusively formed (**24a'**/**24a** : 97/3) starting from the (Z)-unsaturated ester **23b** (Scheme 11).

Scheme 11



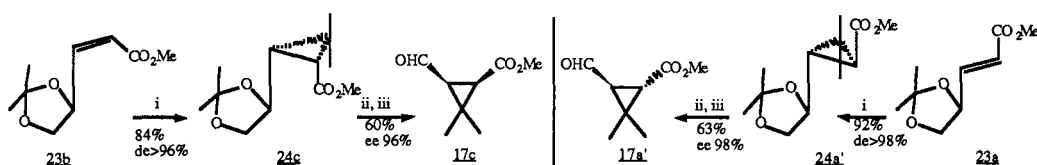
(i) 1.5 equiv. Ph₃P=C(Me)₂, LiI, THF, 0°C, 1h then 20°C, 1h - (ii) 4 equiv. 2N aq. HClO₄, THF, 20°C, 6h - (iii) 1.5 equiv. NaIO₄, MeOH, Phosphate buffer pH 7.2, 20°C, 1h.

These results clearly show that the (1R)-*cis*-series precursor of deltamethrin **1c** is not directly accessible from isopropylidetriphenylphosphorane. It was therefore interesting to test another reagent able to transfer an isopropylidene moiety with conservation of the stereochemistry of the starting α,β -unsaturated ester.

On the basis of Corey's original work ^{14a} and of the results already obtained in our laboratory,^{11b} we expected that isopropylidenediphenylsulfurane could fulfil these requirements.

We effectively found that the unsaturated esters **23b** and **23a** derived from D-glyceraldehyde and possessing the (*Z*)- and (*E*)-stereochemistry react with isopropylidenediphenylsulfurane (1.5 equiv., DME, -78°C, 0.3h then -60°C to -50°C, 0.7h, then -50°C to 20°C) and produce respectively the *cis*-cyclopropyl ester **24c** (84% yield, de : >96%) and the *trans*-cyclopropyl ester **24a'** (92% yield, de : >98%) in very high yield and with very high stereoselectivity (Scheme 12).

Scheme 12



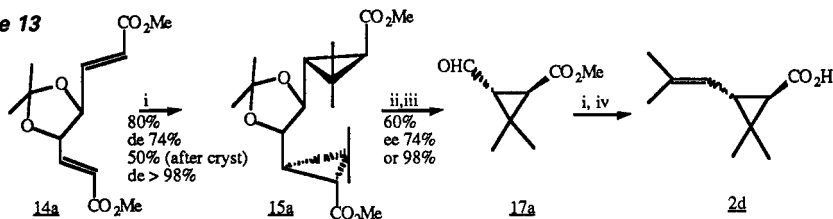
(i) 1.5 equiv. $\text{Ph}_2\text{S}=\text{C}(\text{Me})_2$, DME, -78°C, 0.2 h then -78°C to -50°C, 0.7 h then -50°C to 20°C, 0.3 h (ii) 4 equiv. 2N aq. HClO_4 , THF, 20°C, 6h - (iii) 1.5 equiv. NaIO_4 , MeOH, Phosphate buffer pH 7.2, 20°C, 1h.

These results are in sharp contrast with those obtained with isopropylidetriphenylphosphorane and reported above. (i) The reaction with isopropylidenediphenylsulfurane is almost completely stereospecific as far as the relative stereochemistry is concerned [*trans*-**24a'** from (*E*)-**23a** ; *cis*-**24c** from (*Z*)-**23b**] whereas it is almost completely stereoselective with its phosphorus analogue [*trans*-**24a** and **24a'** from (*E*)-**23a** and (*Z*)-**23b** respectively]. (ii) The stereochemistry of the β -carbon of the adduct **24** is independent of the stereochemistry of the carbon-carbon double bond of **23** when the sulfur ylide is used (always (*S*)) whereas it is completely dependent on it when its phosphorus analogue is involved ((*R*) from (*E*)-**23a** and (*S*)- from (*Z*)-**23b**).

These unexpected results are not yet rationalized, although they fit in the series of results already published ¹⁷ and collected in the scheme 10, but as far as the synthetic aspect is concerned, it is now possible after hydrolysis of the dioxolanes **24** (4 equiv. 2N aq. HClO_4 , THF, 20°C, 6h) and further cleavage of the corresponding diols **16** (NaIO_4 , MeOH, phosphate buffer, pH : 7.2, 20°C, 1h) to the hemicaraldehydes **17** to devise straightforward enantioselective routes to natural pyrethrins and to deltamethrin from natural tartaric acid and from natural D-glyceraldehyde by the proper choice of the isopropylidene transferring reagent (phosphorous or sulfur ylides respectively).

With these results in hand we performed the cyclopropanation of the chiral diesters **14**. In order to achieve the synthesis of the 1R-*trans*-chrysanthemic acid **2d**, the chiral *trans*-diester **14a** has been reacted with 2.5 equiv. of isopropylidetriphenylphosphorane (generated from isopropyl triphenylphosphonium iodide and *n*-butyl lithium in THF-hexane), the reaction being performed at 0°C for 1h then at 20°C for one more hour (Scheme 13). Typically, the diadduct **15** is obtained in 80% yield as a 87/13 mixture of the **15a/15a'** stereoisomers from which the major one's (**15a** m.p. : 123°C, cyclohexane) has been isolated in 50% overall yield after one crystallization. The diadduct **15a** was in turn transformed to (1R)-*trans*-hemicarbaldehyde ⁹ **17a** after hydrolysis of the dioxolane moiety (2N aq. HClO_4 , THF, 20°C, 6h, 98% yield) and cleavage of the resulting diol **16a** (1.5 equiv. NaIO_4 , MeOH, phosphate buffer pH : 7.2, 20°C, 1h) leading to the (1R)-*trans*-hemicaraldehyde **17a** in 68% yield and high enantiomeric excess (ee : 98%). The synthesis of the natural methyl *trans*-chrysanthemate **2d** from **17a** is straightforward and has already been described ¹ (Scheme 13).

Scheme 13



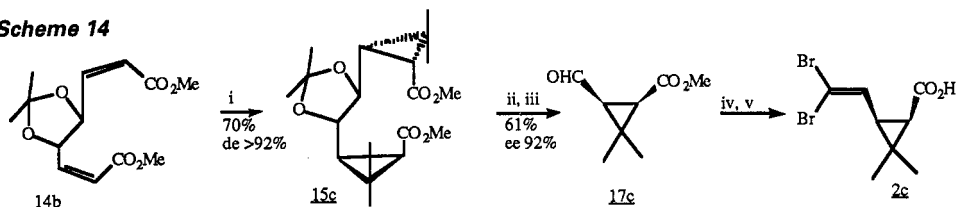
(i) 2.5 equiv. $\text{Ph}_3\text{P}=\text{C}(\text{Me})_2$, LiI, THF, 0°C , 1h then 20°C , 1h - (ii) 4 equiv. 2N aq. HClO_4 , THF, 20°C , 6h - (iii) 1.5 equiv. NaIO_4 , MeOH, Phosphate buffer pH 7.2, 20°C , 1h - (iv) KOH, MeOH then aq. HCl.

The conditions described above for the cyclopropanation reaction are crucial for its success since for example a mixture of mono-adducts and di-adducts, in which the amount of the diastereoisomer **15a** decreases, is produced if the cyclopropanation reaction is carried out between -78°C and 20°C instead of between 0°C and 20°C .

We then envisage the synthesis of the 1R-*cis*-dibromovinyl chrysanthemic acid which is part of deltamethrin the most potent insecticide known. According to the results reported above, we performed the cyclopropanation reaction on the (*Z,Z*)-diester **14b** with isopropylidenediphenylsulfurane (3 mol. equiv.) (DME, -78°C , 0.3h then -60°C to -50°C , 0.7h, then -50°C to 20°C) as the cyclopropanating agent. We obtained the diadduct **15c** possessing two *cis*-disubstituted cyclopropane moieties in 70% overall yield almost as a single diastereoisomer (*de* : $>92\%$) (Scheme 14). This adduct has been in turn transformed to the desired (1R)-*cis*-hemicarinaldehyde **17c** precursor of deltamethrin ¹ after hydrolysis of the dioxolane moiety (2N aq. HClO_4 THF, 20°C , 6h, 98% yield) and cleavage of the resulting diol (1.5 equiv. NaIO_4 , MeOH, phosphate buffer pH : 7.2, 20°C , 1h, 62% yield, *ee* : 92 %).

These results are remarkable since they involve two cyclopropanation reactions, each one occurring in particularly high yield ($> 85\%$) and with high stereochemical control (*de* : $> 96\%$).

Scheme 14



(i) 3 equiv. $\text{Ph}_2\text{S}=\text{C}(\text{Me})_2$, DME, -78°C , 0.3 h then -60°C to -50°C , 0.7 h then -50°C to 20°C - (ii) 6 equiv. 2N aq. HClO_4 , THF, 20°C , 6h - (iii) 1.5 equiv. NaIO_4 , MeOH, Phosphate buffer pH 7.2, 20°C , 1h - (i) 4 $\text{CBr}_4\text{-PPh}_3$, CH_2Cl_2 - (v) KOH, MeOH then aq. HCl - (vi) 2.5 equiv. $(\text{EtO})_2\text{P}(\text{O})\text{CHNaCO}_2\text{Me}$, DME, -78°C then -78°C to $+20^\circ\text{C}$, 4h.

In conclusion, we have devised new stereoselective routes to *cis*- and *trans*-chrysanthemic acid and to related compounds. The second approach we have described is particularly interesting since tartaric acid is readily removed as mixed sodium and potassium tartrate from champagne wine and then transformed by using the route depicted in the Scheme 14 to deltamethrin, the most potent insecticide known. This powerful insecticide is in turn able to protect the winery against the predation of insects. Since only thirty seven grams of such compound are sufficient for that purpose and around ten grams of tartrates have to be removed during the production of champagne wine, we can roughly assume that one hectare of winery in the champagne region can allow the synthesis of enough deltamethrin to protect it for at least a further hundred year.

REFERENCES

- (a) Elliot M. and Janes N.F., *Chem. Soc. Rev.*, **7**, 473 (1978). (b) Arlt D., Jautelat M. and Lantzsch R., *Angew. Chem. int. ed.*, **20**, 703 (1981). (c) Naumann K., *Chemie der Pflanzenschutz- und Schädlingsbekämpfungsmittel*, by Springer-Verlag, Ed., Heidelberg, 1981, ISBN 3-540-10452-6. (d) Deltaméthrine, Roussel-Uclaf, 1982, ISBN 2-904125-00-0. (e) A. R. Justum, R.F.S. Gordon and C.N.E. Ruscoe, *Proceedings 1986, British Corp. Protection Conference*, **1**, 97 (1986). (f) E. Mc Donald, N. Punja and A.R. Justum, *Proceedings 1986, British Corp. Protection Conference*, **1**, 199 (1986).

2. Kiwus R., Schwarz W., Rosnagel I. and Musso H., *Chem. Ber.* **120**, 435 (1987).
3. (a) This compound has been already synthesized by Nozaki ^{3b}. However the problem lies in the preparation of 3-bromo -2,2,5,5-tetramethylcyclohexa-1,3-dione **5** which we were unable to perform selectively under various experimental conditions. Under the conditions cited by Nozaki ^{3b} (ref. to Stetter's work) and using NBS, the bromination is very slow and provides a mixture of products from which the desired pure product **5** can be isolated in low yield by crystallization. (b) Okada T., Kamogawa K., Kawanisi M. and Nozaki H., *Bull. Chem. Soc. Jpn.*, **43**, 2908 (1970).
4. (a) Grob C.A., *Angew. Chem. Int. Ed.*, **8**, 535 (1969) and references cited. (b) Carruthers W., *Some Modern Methods of Organic Chemistry Synthesis*, Cambridge University Press, Cambridge, 107 (1971).
5. House H.O., *Modern Synthetic Reactions*, W.A. Benjamin Inc. Menlo Park., 2nd Ed., (1972).
6. (a) Eliel E.L. and Senda Y., *Tetrahedron*, **26**, 2411 (1970), Report 61. (b) Wigfield D.C., *Tetrahedron* **35**, 449 (1979).
7. (a) Luche J-L., *J. Am. Chem. Soc.*, **100**, 2226 (1978). (b) Luche J-L., *J. Am. Chem. Soc.*, **103**, 5454 (1981) and references cited.
8. (a) Wender P.A., *Heterocycles*, **25**, 263 (1987). (b) Declercq P., University of Ghent, private communication.
9. Martel J., Roussel Uclaf, Fr. Appl. 159,066 (1968), Ger. Offen. 1,966,839 (1984), *Chem. Abst.*, **81**, 135530 (1974).
10. (a) Suzukamo G. and Yasuda M., *Bunri-Gijutsu*, **16**, 345 (1986) (b) Sakita Y. and Suzukamo G., *Chem. Lett.*, **621** (1986) and references cited.
11. (a) Devos M-J., Denis J.N., Krief A., *Tetrahedron Lett.*, 1847 (1978). (b) Devos M-J. and Krief A., *Tetrahedron Lett.*, 1891 (1979).
12. (a) Devos M-J., Hevesi L., Bayet P., Krief A., *Tetrahedron Lett.*, 3911 (1976). (b) Devos M-J. and Krief A., *Tetrahedron Lett.*, **24**, 103 (1983) and references cited. (c) Devos M-J. PhD Thesis Namur (1981) and unpublished results from our laboratory.
13. (a) Devos M-J., Hevesi L., Mathy P., Sevrin M., Chaboteaux G. and Krief A., *Organic Synthesis : an interdisciplinary challenge*, Blackwell, 123 (1985). (b) Chaboteaux G. and Krief A., *Bull. Soc. Chim. Belg.*, **94**, 495 (1985). (c) Krief A. and Devos M-J., *Tetrahedron Lett.*, **26**, 6115 (1985). (d) Krief A., Hevesi L., Chaboteaux G., Mathy P., Sevrin M. and Devos M-J., *J. Chem. Soc. Chem. Commun.*, 1693 (1985).
14. (a) Corey E.J. and Jautelat M., *J. Amer. Chem. Soc.*, **89**, 3913 (1967). (b) Sevrin M., Hevesi L., Bayet P. and Krief A., *Tetrahedron Lett.*, 3915 (1976).
15. (a) Debost J.L., Gelas J. and Horton D., *J. Org. Chem.*, **48**, 1381 (1983). (b) Carmack M. and Kelly O.J., *J. Org. Chem.*, **33**, 2971 (1968).
16. (a) Wadsworth W.S., *Organic Reactions*, **25**, 73 (1977). (b) Minami N., Ko S.S., Kishi Y., *J. Am. Chem. Soc.*, **104**, (1982). (c) Jurczak J., Pikul S. and Bauer T., *Tetrahedron*, **42**, 447 (1986).
17. (a) Matsunaga H., Sakamaki T., Nagoka H. and Yamada Y., *Tetrahedron Lett.*, **23**, 3009 (1983). (b) Trost B.M. and Mignani S.M., *Tetrahedron Lett.*, **35**, 4137 (1986). (c) Mulzer J. and Kappert M., *Tetrahedron Lett.*, **26**, 1631 (1985). (d) Katsuki T., Lee A.W.M., Ma P., Machin V.S., Masamune S., Sharpless K.B., Tuddenham D. and Walker F.J., *J. Org. Chem.*, **47**, 1378 (1982). (e) Mulzer J. and Kappert M., *Angew. Chem. Int. Ed.*, **22**, 63 (1983). (f) Yamamoto Y., Nishii S. and Ibuka T., *J.C.S. Chem. Commun.*, 464 (1987). (g) Roush W.R. and Lesur B.M., *Tetrahedron Lett.*, **23**, 2231 (1983). (h) Stork G. and Kahn M., *Tetrahedron Lett.*, **24**, 3951 (1983).